# The three R's of a good structure, Resolution, Refinement

## Eddie Snell

# The structure from a crystal is the structure of the macromolecule?



Right?

Well yes, but ....

It's the average structure.

It's averaged over many individual macromolecules.

And it's averaged over time.

'Look, Watkins - I've invented a new prehistoric creature!'

Most importantly it's a MODEL that best explains the data

## **A Crystal**

Some sites occupied by macromolecules Dynamics going on

Others not

A regular lattice

### Many frozen crystals

Even when the crystal is cryocooled, dynamic motion is 'frozen in' and chemical processes can be ongoing

# How do we store structures?

Structures are deposited in the protein data bank or PDB (which also includes other biological macromolecules)

http://www.rcsb.org



A tutorial is available at http://www.rcsb.org/pdb/tutorials/tutorial.html

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### The Unit Cell





### The Essence of Structural Crystallography

 There are known knowns. These are things we know that we know. There are known unknowns. That is to say, there are things that we know we don't know. But there are also unknown unknowns. There are things we don't know we don't know.

![](_page_10_Picture_2.jpeg)

• Donald Rumsfeld, Feb 12<sup>th</sup> 2002.

Shamelessly copied from a slide by Ted Baker

### The three R's of a good structure

- A structure is a model that best represents the measured data.
- Think about what you are measuring:
  - The data is an average taken over many macromolecules. For example, a 100 µm<sup>3</sup> crystal produced from a macromolecule that has a typical size of 200 Å on edge will consist of ~ 5,000 molecules on edge or 125,000,000,000 molecules in total.
  - The data is not static, it represents an average of those molecules over time.
  - The data is dynamic. X-rays cause chemical changes which can also be captured over time.
- Given these, how do we get a good structure.
  How do we know when we have a good structure?

### Known knowns - we know what to expect

![](_page_12_Figure_1.jpeg)

![](_page_12_Figure_2.jpeg)

Structure Validation by C $\alpha$  Geometry:  $\phi$ ,  $\psi$  and C $\beta$  Deviation PROTEINS: Structure, Function, and Genetics 50:437–450 (2003)

Simon C. Lovell, Ian W. Davis, W. Bryan Arendall III, Paul I. W. de Bakker, J. Michael Word, Michael G. Prisant, Jane S. Richardson, and David C. Richardson

The dihedral angles in the main chain have allowed and disallowed regions that are well known – developed by Gopalasamudram Narayana Ramachandran and called the Ramachandren plot. Available as part of several software packages.

### Known knowns – we know what to expect

![](_page_13_Picture_1.jpeg)

### Known knowns - we know what to expect

![](_page_14_Figure_1.jpeg)

![](_page_14_Figure_2.jpeg)

We can look for deviations in bond chain angles and lengths from previous data, e.g. Procheck <u>http://www.biochem.ucl.ac.uk/~roman/procheck/procheck.html</u>

### Known knowns - we know what to expect

![](_page_15_Figure_1.jpeg)

![](_page_15_Figure_2.jpeg)

Similarly we can look for deviations in properties and geometry from previous data, e.g. Procheck <u>http://www.biochem.ucl.ac.uk/~roman/procheck/procheck.html</u>

Some of these checks are incorporated within building and display programs, e.g. Coot

![](_page_16_Figure_1.jpeg)

http://www.ysbl.york.ac.uk/~emsley/coot/

The validation option gives many options on how to check your structure against the data and against stereochemical restraints and previous structural information

![](_page_17_Figure_1.jpeg)

# Known knowns – we know what not to expect

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#### QUICK GUIDE TO CHARACTERISTICS OF METAL SITES IN PROTEINS

metal	Na sodium	Mg ** magnesium	K potassium	Ca calcium	Mn manganese
atomic no. (= no. of electrons)	11	12	19	20	25
usual ion	Na+	Mg <sup>2+</sup>	K+	Ca <sup>2+</sup>	Mn <sup>2+</sup> (Mn <sup>3+</sup> )
usual donor atoms <sup>1</sup> (for more detail see )	Om.chain Oofasp,glu	O of asp,glu O m.chain	O m.chain O of asp,glu,	O of asp,glu, O m.chain	O of asp,glu N of his
other donors sometimes found (and occasionally found)	O of ser,thr O of asn,gln (N of his)	O of ser,thr O of asn,gln N of his	O of ser,thr O of asn,gln (N of his)	O of asn,gln O of ser,thr (N of his)	Om.chain Oofasn,gln Oofser,thr
usual coordination number(s) <sup>2</sup>	5,6	6	5,6	6	5,6
other coordination numbers <sup>3</sup>	4,7	3,4,5	4,7,8	4,5,7,8	4,7
typical <sup>4</sup> distance Å) M-O <u>more info</u> M-N M-S	2.35-2.45	2.05-2.15 	2.75-2.85 ?+ 	2.35-2.45	2.15-2.20 2.21 2.35
relative abundance <sup>5</sup> in PDB	96	177	72	358	109
link to lists of examples <sup>6</sup>	<u>Na groups</u>	Mg groups	K groups	Ca groups	Mn groups
** chlorophyll groups, containing	Mg, are common, k	out not included he	re		•
ine 😜 Internet 🔍 100%					

There are many sources of typical bond distances, e.g. Metal Coordination Sites in Proteins at:

#### http://tanna.bch.ed.ac.uk/

# Known unknowns – we know when we have something we don't know

![](_page_19_Picture_1.jpeg)

X-rays are diffracted due to interactions with electrons in the atoms.

The data we produce is a map of electron density.

# Known unknowns – we know when we have something we don't know

![](_page_20_Figure_1.jpeg)

If we plot electron density maps as:

(Two times the observed data – the calculated data) in blue (2Fo-Fc)

And (the observed data – the calculated data) and color this according to negative (red) and positive (green) (Fo-Fc)

We know we have something we don't know.

# Known unknowns – we know when we have something we don't know

![](_page_21_Figure_1.jpeg)

If we model the correct atoms, i.e. a zinc finger motif then the maps tells us that we got it correct.

We can adjust our model to add the 'known something we didn't know'.

### Unknown unknowns – The R-factor

 $I_{hkl} \propto |F(hkl)|^2$ 

The X-ray intensity for a particular reflection (hkl),  $I_{hkl}$ , is measured,  $F_{obs}$  is related to the measured I.

The X-ray intensity for a particular reflection (hkl) can also be calculated from the model,  $I_{calc}$ , is calculated,  $F_{calc}$  is related to the calculated I and the model.

$$R = \frac{\sum ||F_{obs}| - |F_{calc}||}{\sum |F_{obs}|}$$

The R factor – small is good.

A small R-factor indicates minimal differences between the electron density calculated for the model and that calculated from the observed data.

### The R-factor – What does it mean?

$$R = \frac{\sum ||F_{obs}| - |F_{calc}||}{\sum |F_{obs}|}$$

If the observed measurement is in complete agreement with the calculated measurement then R will equal zero.

If there is disagreement then R will be finite. R is expressed as a percentage and is typically a little better than ten times the resolution for a good structure.

### The R<sub>free</sub>-factor – What does it mean?

$$R = \frac{\sum ||F_{obs}| - |F_{calc}||}{\sum |F_{obs}|}$$

Some 5-10% of the reflections are not used to calculate the model. These are used to calculate an  $R_{free}$ .

When the model is improving, i.e. it is accurately explaining the data and both the R and R<sub>free</sub> should reduce during refinement. Once the Rfree stops reducing the model is being overfitted to the data – it is losing accuracy.

![](_page_25_Figure_0.jpeg)

The B factor (or atomic displacement factor) describes how the electron is spread out in space. A high B factor would indicate a high degree of uncertainty in the atomic position and a potential warning sign.

# Very low Resolution

## Low Resolution

# Medium Resolution

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#### High Resolut

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## Very-high Resolution

TOTEL PATTONIA

![](_page_31_Figure_0.jpeg)

The amino acids that form the building blocks of biological molecules

Let's pay particular attention to one of them and the concept of resolution

Borrowed from http://andersenlab.chem.washington.edu/CSDb/pics/amino-acids.png

![](_page_32_Figure_0.jpeg)

At 2.50Å, the hole in the middle of the ring is gone.

![](_page_32_Figure_2.jpeg)

#### Quality (Resolution)

#### Low resolution

![](_page_32_Picture_5.jpeg)

#### High resolution

![](_page_32_Picture_7.jpeg)

![](_page_32_Figure_8.jpeg)

![](_page_32_Figure_9.jpeg)

At 1.25 Å, atoms are mostly resolved but hydrogen atoms are much less evident

Sean Parkin. UK

### Where does Reality come in?

- The model is accurate if it can be validated. The higher the resolution then the more precise the detail in the model.
- Remember, it is only a model that explains the data and not data that explains the model.

### Examples of Publically available software

#### Refinement:

- Phenix, a software suite for doing just about everything with data but displaying the structure: <u>http://www.phenix-online.org/</u>
- CCP4, similar to Phenix but covering a broarder area of application and including validation and display components:<u>http://www.ccp4.ac.uk/main.html</u>
- CNS, Crystallography and NMR system: <a href="http://cns-online.org/v1.21/">http://cns-online.org/v1.21/</a>

#### Validation:

- Excellent tutorial on validation: <u>http://xray.bmc.uu.se/gerard/embo2001/modval/index.html</u>
- Procheck is a good validation example: <u>http://www.biochem.ucl.ac.uk/~roman/procheck/procheck.html</u> and is incorporated with CCP4.
- MolProbity allows both validation and fixing the model: <u>http://molprobity.biochem.duke.edu/</u>
- Direct valisation through the PDB: <a href="http://sw-tools.pdb.org/apps/VAL/index.html">http://sw-tools.pdb.org/apps/VAL/index.html</a>

Display:

• Coot (also includes validation routines): <u>http://www.ysbl.york.ac.uk/~emsley/coot/</u>